

# Synthesis of N-Substituted 4-(2-Oxoalkyl)-1,4-dihydronicotinates and Their Inverse Electron Demand Diels-Alder Reaction with 3,4-Dichlorothiophene 1,1-Dioxide

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Received July 2, 1986

N-Substituted 4-(2-oxoalkyl)-1,4-dihydronicotinates **4** were prepared in high yields by the reaction of quaternized nicotinium salts with silyl enol ethers. These 1,4-dihydronicotinates acted as dienophiles in Diels-Alder reactions with the electron-deficient 3,4-dichlorothiophene 1,1-dioxide. The cycloaddition proceeded regio- and stereoselectively in good yield to give 1,4,4a,8a-tetrahydroquinolines **8**.

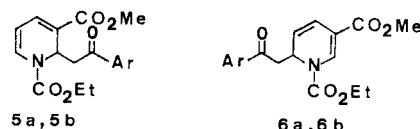
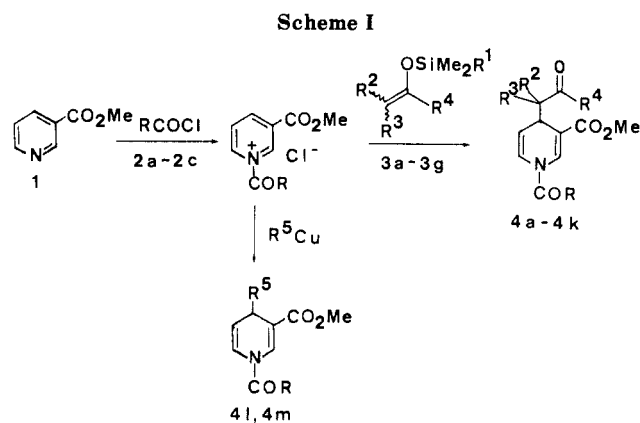
Dihydropyridines have been extensively studied,<sup>1</sup> and current interest in these compounds includes NADH, synthesis of NADH model compounds, and the mechanism of reduction by these compounds.<sup>2</sup> 1,2-Dihydropyridines have been utilized as dienes in intra- or intermolecular Diels-Alder reactions to afford isoquinuclidines.<sup>3,4</sup> In contrast, 1,4-dihydropyridines have not been utilized as dienophiles, although a Diels-Alder reaction is very useful for the construction of heterocycles with high stereoselectivity. The absence of such reactions may be due to the electron-rich nature of the reagent and the thermal instability of some dihydropyridines. However, 1,4-dihydronicotinates bearing an electron-withdrawing substituent should be stable. In this paper we describe the synthesis of 1,4-dihydronicotinates **4a-m**, their inverse electron demand Diels-Alder reactions with 3,4-dichlorothiophene 1,1-dioxide (**7**), and some stereochemical features of the addition.

**Synthesis of Methyl 4-(2-Oxoalkyl)-1,4-dihydronicotinates.** We have reported the regioselective synthesis of 1-acyl-4-(2-oxoalkyl)-1,4-dihydropyridines<sup>5</sup> and other 4-substituted 1,4-dihydropyridines<sup>6</sup> by reaction of quaternized pyridines with silyl enol ethers. We have now used this reaction to obtain methyl 4-(2-oxoalkyl)-1,4-dihydronicotinates **4a-k** in yields of 58–97% (Table I). When we used the trimethylsilyl enol ethers of acetophenones (**3a,b**, R<sup>1</sup> = Me), 1,2- (**5**) and 1,6-dihydronicotinates (**6**) were obtained as byproducts (4/5/6 = 5:1:1) (Scheme I). Use of *tert*-butyldimethylsilyl enol ethers **3a** and **3b** (R<sup>1</sup> = *t*-Bu) improved this ratio to 10:1:1.<sup>5</sup> More severe reaction conditions were required for preparation of **4i-k** since the formation of the quaternized nicotinium salts was sluggish because of the lower reactivity of di-

Table I. Synthesis of 1,4-Dihydronicotinates **4**

<b>4</b>	R	R <sup>2</sup>	R <sup>3</sup>	R <sup>4</sup>	R <sup>5</sup>	total yield, %
<b>4a</b>	OEt	H	H	Ph		63 <sup>a,b</sup>
<b>4b</b>	OEt	H	H	<i>p</i> -MeOC <sub>6</sub> H <sub>4</sub>		58 <sup>a,c</sup>
<b>4c</b>	OEt	H	Ph	OMe		85
<b>4d</b>	OEt	Me	Me	OMe		61
<b>4e</b>	OMe	H	Me	Et		69
<b>4f</b>	OMe	H	Me	Ph		96
<b>4g</b>	OMe	Me	Me	Me		93
<b>4h</b>	Me	Me	Me	Me		76
<b>4i</b>	NMe <sub>2</sub>	Me	Me	Me		63
<b>4j</b>	NMe <sub>2</sub>	H	Ph	OMe		97
<b>4k</b>	NMe <sub>2</sub>	Me	Me	OMe		60
<b>4l</b>	OMe				Me	76
<b>4m</b>	OMe				<i>p</i> -Tolyl	71

<sup>a</sup> *tert*-Butyldimethylsilyl enol ether was used. <sup>b</sup> Product ratio 1,4/1,2/1,6 = 10:1:1. <sup>c</sup> 1,4, 47%; 1,2 + 1,6, 11%.



2a: R=MeO  
2b: R=EtO  
2c: R=Me<sub>2</sub>N

methylcarbamoyl chloride.<sup>7</sup> 4-Alkyl-1,4-dihydronicotinates **4l,m** were synthesized by reaction of the nicotinium salt with methylcopper or *p*-tolylcopper prepared from methyllithium or *p*-tolylolithium and 1 equiv of copper(I) iodide.<sup>8</sup>

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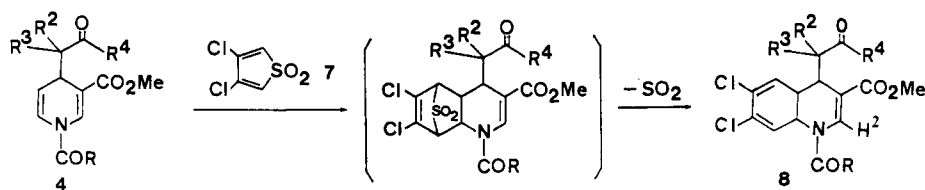
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Scheme II

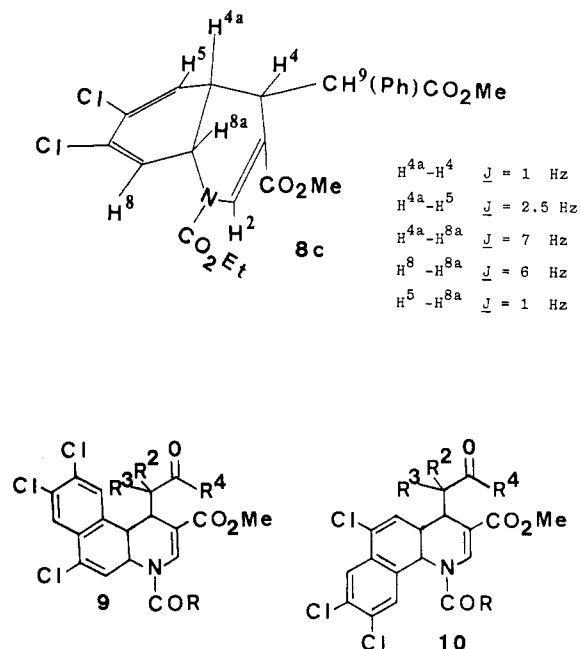


**Reaction of 4 with Dienes.** The reactions of **4g** with three dienes were carried out to explore its reactivity as a dienophile. When 2,3-dimethyl-1,3-butadiene was heated with **4g** in benzene-*d*<sub>6</sub> to 60 °C for 33 h, **4g** was recovered in 90% yield. No reaction occurred on heating **4g** with a large excess of 1-methoxy-3-(trimethylsiloxy)-1,3-butadiene (Danishefsky's diene<sup>9</sup>) at 100 °C for 7 h in toluene. However, when **4g** was heated with 2 equiv of 3,4-dichlorothiophene 1,1-dioxide (**7**)<sup>10</sup> in refluxing toluene for 12 h, a rather complex mixture was obtained from which the expected adduct **8g** was isolated in 15% yield. Other products included recovered **4g** (4%), the dimer of **7**, and methyl nicotinate (**1**, 36%), probably formed by the decomposition of **4g**. The fact that **8g** was obtained encouraged us to explore reactions of **7** with other 1,4-dihydropyridinates with different substituents at the C-4 position. Although **4a** gave the same results as **4g** (**8a**/1/**4a** = 21:38:20), no **1** was obtained in the reactions of **4c** and **4d**, showing the different effects of the β-(alkoxycarbonyl)alkyl group and the 2-oxoalkyl group at the C-4 position of **4**.

The structures **8** were determined by <sup>1</sup>H NMR, MS, and elemental analyses to be the 1:1 adducts of **4** and **7** with the elimination of SO<sub>2</sub>. When we carried out the reaction with **4a,d,g**, the diastereomerically pure **4c**, only one isomer of the corresponding **8** was obtained. Thus the reaction proceeded regio- and stereoselectively (syn and anti). One notable feature of the <sup>1</sup>H NMR of **10** is a singlet at low field (δ 8.1–8.3 for R = OMe, OEt; δ 7.2–7.7 for R = NMe<sub>2</sub>). Such a signal is characteristic of a β-proton in an α,β-unsaturated ester, that is, H<sub>2</sub> in the adduct **8**. Consequently, cycloaddition must have taken place at the 5,6-double bond in **4** to give **8** as shown in Scheme II. Since the <sup>1</sup>H NMR patterns of **8** were quite similar, the structure of a diastereomerically pure sample of **8c** was determined precisely in order to elucidate the stereochemistry of the cycloaddition. In order to determine the relative stereochemistry at C<sub>4a</sub>–C<sub>8a</sub> and C<sub>4</sub>–C<sub>4a</sub>, the relationships between protons were determined by decoupling experiments, assuming that the lowest field proton peak (δ 3.87) of the four sp<sup>3</sup> ones belonged to H<sub>8a</sub> adjacent to the nitrogen atom. The results are shown in Scheme III.

We first examined Dreiding models of the ring junction and found that in cis-fused **8** there are two conformations in which the dihedral angle between H<sub>4a</sub>–H<sub>8a</sub> is ca. 60°. However, in trans-fused **8** there is only one rigid conformation, with a dihedral angle of ca. 180°. The coupling constant *J*<sub>4a–8a</sub> of **8c** was 7 Hz, indicating that the ring junction is cis. In one of the two conformations of cis-fused **8**, the dihedral angles between H<sub>4a</sub>–H<sub>5</sub> and H<sub>8</sub>–H<sub>8a</sub> are ca. 30° and ca. 90°, respectively; in the other conformer

Scheme III



the magnitudes of the angles are reversed. In **8c**, *J*<sub>4a–5</sub> is 2.5 Hz and *J*<sub>8–8a</sub> is 6 Hz, thus the conformation shown in Scheme III is preferred. We observed 18% NOE at H<sub>5</sub> (δ 5.57) when H<sub>4</sub> and H<sub>9</sub> (δ 3.53–3.63) were irradiated, with less NOE at H<sub>4a</sub> (5–9%; δ 2.92). Hence it can be concluded that the diene moiety is anti to the C<sub>4</sub> substituent. The exclusive formation of the cis-anti adduct shows that the reaction proceeded not in a stepwise but in a concerted manner from the less hindered side; a similar stereoselectivity has been reported in the 1,3-dipolar cycloaddition with 1,2-dihydropyridines.<sup>11</sup>

Two problems were examined in efforts to improve the yield of **8**. One was the secondary Diels–Alder reaction of **8** with **7** to give **9** and/or **10**. In a reaction of **4d** with **7**, **9d** was obtained in 8% yield. The structures of **9** and **10** were assigned by <sup>1</sup>H NMR. The use of a large excess and stepwise addition of **7** could be the cause for the secondary reaction, and better yields of **8c** and **8d** were obtained when 5 equiv of **7** was added to a concentrated toluene solution of **4c** or **4d** all at once and the mixture heated to reflux. The desired end point of the reaction was determined by monitoring with TLC.

A second problem was the decomposition of **4** to **1**, which is probably caused by HCl generated in the dimerization of **7**. Actually this decomposition took place only with β-keto substituents (**4a,g**) and not with β-alkoxycarbonyl substituents (**4c,d**). Because of the sensitivity of **4** to HCl, we employed acetyl and dimethylcarbamoyl chlorides instead of ethoxycarbonyl chloride as acylating reagents. However, in the reaction of the *N*-acetyl compound **4h** with 2 equiv of **7**, only 3% yield of **8b** was obtained and 57% of methyl nicotinate (**1**) was recovered. Likewise, the *N*-carbamoyl compound **4k** reacted with 3 equiv of **7** to

(8) From pyridinium salt with alkyl- or arylcopper or alkyl- or arylcopper–BF<sub>3</sub> complex. Akiba, K.-y.; Iseki, Y.; Wada, M. *Bull. Chem. Soc. Jpn.* 1984, 57, 1994. From pyridinium salt with alkyl or aryl Grignard reagent–5% cuprous iodide, see: Comins, D. L.; Abdullah, A. H. *J. Org. Chem.* 1982, 47, 4315. Comins, D. L.; Stroud, E. D.; Herrick, J. J. *Heterocycles* 1984, 22, 151.

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Table II. Synthesis of 1,4,4a,8a-Tetrahydroquinoline<sup>a</sup>

entry	8	R	R <sup>2</sup>	R <sup>3</sup>	R <sup>4</sup>	R <sup>5</sup>	base	yield recov, %		
								8	4	1
1	8a	OEt	H	H	Ph			21	20	38
2	8a	OEt	H	H	Ph		<i>b</i>	61	13	
3	8b	OEt	H	H	<i>p</i> -MeOC <sub>6</sub> H <sub>4</sub>		<i>b</i>	60	0	
4	8c	OEt	H	Ph	OMe			52	11	
5	8d	OEt	Me	Me	OMe			36	60	
6	8e	OMe	H	Me	Et		<i>b</i>	55	38	
7	8f	OMe	H	Me	Ph		<i>b</i>	31	38	
8	8g	OMe	Me	Me	Me		<i>b</i>	22	65	
9	8i	NMe <sub>2</sub>	Me	Me	Me		<i>b</i>	26	40	
10	8j	NMe <sub>2</sub>	H	Ph	OMe			54	29	
11	8k	NMe <sub>2</sub>	Me	Me	OMe			36	39	
12	8l	OMe				Me		70	0	
13	8m	OMe				<i>p</i> -Tolyl		65	0	

<sup>a</sup> Entries 1, 4, 5, and 10–13, 4/7 = 1:5; entries 2, 3, and 6–9, 4/7/base = 1:10:15. <sup>b</sup> 2,6-Lutidine was used as the base.

give only 5% yield of **8k** with 60% recovery of **1**. Therefore we tried to avoid this decomposition by the addition of a base. 1,8-Diazabicyclo[5.4.0]undec-7-ene (DBU), 1,8-bis(dimethylamino)naphthalene (proton sponge), or pyridine reacted with **7** immediately at room temperature to form black tars. In contrast, 2,2,6,6-tetramethylpiperidine, 1,2,2,6,6-pentamethylpiperidine, 2,6-lutidine, and 4-methyl-2,6-ditert-butylpyridine did not react as rapidly with **7**. And we carried out the reaction of **4g** with 5 equiv of **7** in the presence of these sterically hindered bases. The best yield of **8g** (22%) was obtained by use of 2,6-lutidine; 2,2,6,6-tetramethylpiperidine and 1,2,2,6,6-pentamethylpiperidine gave only half as much. A complex reaction occurred with 4-methyl-2,6-di-*tert*-butylpyridine, and no **8g** was obtained. Therefore we carried out the reaction of other **4** with 10 equiv of **7** in the presence of 15 equiv of 2,6-lutidine and obtained still better yields of **8** (Table II). The effect of the base is shown by entries 1 and 2.

In summary, Diels–Alder adducts **8** were obtained in good yields by using excess 2,6-lutidine to trap hydrogen chloride generated in situ. Table II shows that the yield of **8** decreased as the steric hindrance of **4** increased (entries 5,8,9,11), as might be expected because the Diels–Alder reaction is quite susceptible to steric hindrance.

### Experimental Section

Melting points were taken on a micro melting point apparatus and are uncorrected. IR spectra were obtained on a Hitachi 215 spectrometer. <sup>1</sup>H and <sup>13</sup>C NMR spectra were obtained on a Hitachi R-90H spectrometer. Mass spectra were recorded on a Hitachi RMU-6L spectrometer. Flash column chromatography was carried out on Merck silica gel 60, 230–400 mesh. Thin-layer chromatography (TLC) was performed with Merck silica gel GF-254 plates. 1-Methoxy-3-(trimethylsilyloxy)-1,3-butadiene<sup>9</sup> and 3,4-dichlorothiophene 1,1-dioxide<sup>10</sup> were prepared by the reported method.

**Synthesis of 1-[Alkoxy-carbonyl(or acetyl)]-3-(methoxy-carbonyl)-4-(2-oxoalkyl)-1,4-dihydropyridines (4a–h) with Trimethylsilyl Enol Ethers 3. General Procedure.** Ethyl (or methyl) chloroformate (or acetyl chloride) (7 mmol) was added to a solution of methyl nicotinate (7 mmol) in 10 mL of dichloromethane at 0 °C. Trimethylsilyl enol ether **3** (8 mmol) was added, and the reaction mixture was stirred under nitrogen at room temperature for 10–12 h. The resulting reaction mixture was treated with 5% NaHCO<sub>3</sub> (20 mL), and the product was extracted with ether (25 mL × 3). After drying over anhydrous MgSO<sub>4</sub>, the solvent was evaporated in vacuo. The crude product was purified by flash column chromatography with hexane and ethyl acetate (4:1–2:1) or dichloromethane as eluent to afford **4**.

**1-(Ethoxycarbonyl)-3-(methoxycarbonyl)-4-(2-oxo-2-phenylethyl)-1,4-dihydropyridine (4a):** oil; <sup>1</sup>H NMR (CDCl<sub>3</sub>) δ 1.33 (t, 3 H, *J* = 7 Hz, CO<sub>2</sub>CH<sub>2</sub>CH<sub>3</sub>), 3.02 (dd, 1 H, *J* = 15, 9 Hz, CH<sub>2</sub>COPh), 3.41 (dd, 1 H, *J* = 15, 3 Hz, CH<sub>2</sub>COPh), 3.75 (s,

3 H, CO<sub>2</sub>Me), 3.95 (ddd, 1 H, *J* = 9, 5, 3 Hz, H<sub>4</sub>), 4.30 (q, 2 H, *J* = 7 Hz, CO<sub>2</sub>CH<sub>2</sub>CH<sub>3</sub>), 5.25 (dd, 1 H, *J* = 9, 5 Hz, H<sub>5</sub>), 6.76 (d, 1 H, *J* = 9 Hz, H<sub>6</sub>), 7.3–7.6 (m, 3 H, Ph), 7.8–8.2 (m, 3 H, Ph + H<sub>2</sub>).

**1-(Ethoxycarbonyl)-3-(methoxycarbonyl)-4-[2-(*p*-methoxyphenyl)-2-oxoethyl]-1,4-dihydropyridine (4b):** mp 104–105 °C; <sup>1</sup>H NMR (CDCl<sub>3</sub>) δ 1.33 (t, 3 H, *J* = 7 Hz), 2.95 (dd, 1 H, *J* = 15, 6 Hz), 3.42 (dd, 1 H, *J* = 15, 4 Hz), 3.75 (s, 3 H), 3.84 (s, 3 H), 3.90–4.15 (m, 1 H), 4.28 (q, 2 H, *J* = 7 Hz), 5.24 (dd, 1 H, *J* = 4, 8 Hz), 6.75 (dd, 1 H, *J* = 8, 1 Hz), 6.90 (d, 2 H, *J* = 9 Hz), 7.95 (d, 2 H, *J* = 9 Hz), 7.97 (d, 1 H, *J* = 1 Hz). Anal. Calcd for C<sub>19</sub>H<sub>21</sub>O<sub>6</sub>N: C, 63.50; H, 5.89; N, 3.90. Found: C, 63.35; H, 5.89; N, 4.10.

**1-(Ethoxycarbonyl)-3-(methoxycarbonyl)-4-[1-(methoxy-carbonyl)benzyl]-1,4-dihydropyridine (4c).** The initially eluted diastereomer of **4c** from flash column chromatography (dichloromethane): oil; <sup>1</sup>H NMR (CDCl<sub>3</sub>) δ 1.25 (t, 3 H, *J* = 8 Hz), 3.67 (s, 3 H), 3.79 (s, 3 H), 3.9–4.2 (m, 2 H), 4.17 (q, 2 H, *J* = 8 Hz), 5.30 (dd, 1 H, *J* = 5, 8 Hz), 6.67 (dd, 1 H, *J* = 8, 1 Hz), 6.95–7.4 (m, 5 H), 7.73 (d, 1 H, *J* = 1 Hz). Anal. Calcd for C<sub>19</sub>H<sub>21</sub>O<sub>6</sub>N: C, 63.50; H, 5.89; N, 3.90. Found: C, 63.37; H, 6.03; N, 4.17. Another diastereomer: oil; <sup>1</sup>H NMR (CDCl<sub>3</sub>) δ 1.35 (t, 3 H, *J* = 8 Hz), 3.45 (s, 3 H), 3.62 (s, 3 H), 3.5–4.1 (m, 2 H), 4.31 (q, 2 H, *J* = 8 Hz), 5.15 (dd, 1 H, *J* = 5, 7 Hz), 6.85 (dd, 1 H, *J* = 7, 1 Hz), 6.9–7.4 (m, 5 H), 7.93 (d, 1 H, *J* = 1 Hz).

**1-(Ethoxycarbonyl)-3-(methoxycarbonyl)-[1-(methoxy-carbonyl)-1-methylethyl]-1,4-dihydropyridine (4d):** oil; <sup>1</sup>H NMR (CDCl<sub>3</sub>) δ 0.87 (s, 3 H), 1.09 (s, 3 H), 1.36 (t, 3 H, *J* = 7 Hz), 3.65 (s, 3 H), 3.74 (s, 3 H), 3.80 (d, 1 H, *J* = 6 Hz), 4.33 (q, 2 H, *J* = 7 Hz), 5.07 (dd, 1 H, *J* = 8, 6 Hz), 6.95 (dd, 1 H, *J* = 8, 1 Hz), 8.03 (d, 1 H, *J* = 1 Hz). Anal. Calcd for C<sub>15</sub>H<sub>21</sub>O<sub>6</sub>N: C, 57.87; H, 6.80; N, 4.50. Found: C, 58.00; H, 6.61; N, 4.43.

**1,3-Bis(methoxycarbonyl)-(1-methyl-2-oxobutyl)-1,4-dihydropyridine (4e).** The initially eluted diastereomer of **4e** from flash column chromatography (4:1 hexane/AcOEt): oil; <sup>1</sup>H NMR (CDCl<sub>3</sub>) δ 0.96 (d, 3 H, *J* = 7 Hz), 1.06 (t, 3 H, *J* = 7 Hz), 2.33–3.20 (m, 3 H), 3.77 (s, 3 H), 3.90 (s, 3 H), 3.90–4.10 (m, 1 H), 4.92 (dd, 1 H, *J* = 8, 4 Hz), 6.88 (dd, 1 H, *J* = 8, 1 Hz), 7.97 (d, 1 H, *J* = 1 Hz). Anal. Calcd for C<sub>14</sub>H<sub>19</sub>O<sub>5</sub>N: C, 59.78; H, 6.81; N, 4.98. Found: C, 59.74; H, 7.03; N, 4.78. Another diastereomer: oil; <sup>1</sup>H NMR (CDCl<sub>3</sub>) δ 1.00 (d, 3 H, *J* = 7 Hz), 1.00 (t, 3 H, *J* = 7 Hz), 2.18–3.00 (m, 3 H), 3.73 (s, 3 H), 3.90 (s, 3 H), 3.70–4.10 (m, 1 H), 5.10 (dd, 1 H, *J* = 8, 4 Hz), 6.87 (dd, 1 H, *J* = 8, 1 Hz), 7.90 (d, 1 H, *J* = 1 Hz).

**1,3-Bis(methoxycarbonyl)-4-(1-methyl-2-oxo-2-phenylethyl)-1,4-dihydropyridine (4f).** The initially eluted diastereomer of **4f** from flash column chromatography (4:1 hexane/AcOEt): mp 104–106 °C; <sup>1</sup>H NMR (CDCl<sub>3</sub>) δ 1.07 (d, 3 H, *J* = 7 Hz), 3.84 (s, 3 H), 3.88 (s, 3 H), 3.8–4.1 (m, 2 H), 4.78 (dd, 1 H, *J* = 4, 8 Hz), 6.90 (d, 1 H, *J* = 8 Hz), 7.4–7.6 (m, 3 H), 8.0–8.3 (m, 3 H). Another diastereomer: oil; <sup>1</sup>H NMR (CDCl<sub>3</sub>) δ 1.18 (d, 3 H, *J* = 7 Hz), 3.66 (s, 3 H), 3.84 (s, 3 H), 3.7–4.0 (m, 2 H), 5.22 (dd, 1 H, *J* = 5, 8 Hz), 6.84 (dd, 1 H, *J* = 8, 1 Hz), 7.3–7.6 (m, 3 H), 7.7–7.9 (m, 3 H).

**1,3-Bis(methoxycarbonyl)-4-(1,1-dimethyl-2-oxopropyl)-1,4-dihydropyridine (4g):** mp 88–89 °C; <sup>1</sup>H NMR (CDCl<sub>3</sub>) δ 0.96 (s, 3 H), 1.08 (s, 3 H), 2.18 (s, 3 H), 3.73 (s, 3 H), 3.82 (d, 1

H,  $J = 5$  Hz), 3.90 (s, 3 H), 5.05 (dd, 1 H,  $J = 5, 8$  Hz), 6.95 (dd, 1 H,  $J = 8, 1$  Hz), 8.02 (d, 1 H,  $J = 1$  Hz). Anal. Calcd for  $C_{14}H_{19}O_5N$ : C, 59.77; H, 6.81; N, 4.98. Found: C, 59.94; H, 7.04; N, 4.99.

**1-Acetyl-3-(methoxycarbonyl)-4-(1,1-dimethyl-2-oxopropyl)-1,4-dihydropyridine (4h):** oil;  $^1H$  NMR ( $Me_2SO-d_6$ , 100 °C)  $\delta$  0.87 (s, 3 H), 1.01 (s, 3 H), 2.12 (s, 3 H), 2.30 (s, 3 H), 3.67 (s, 3 H), 3.73 (dd, 1 H,  $J = 6, 1$  Hz), 5.12 (dd, 1 H,  $J = 6, 8$  Hz), 7.06 (dd, 1 H,  $J = 8, 2$  Hz), 7.97 (dd, 1 H,  $J = 2, 1$  Hz).

**Synthesis of 1-[(Dimethylamino)carbonyl]-3-(methoxycarbonyl)-4-(2-oxoalkyl)-1,4-dihydropyridines (4i-k). General Procedure.** A mixture of methyl nicotinate (7 mmol) and dimethylcarbonyl chloride (9 mmol) was heated to 100 °C for 1 h. To this solution was added trimethylsilyl enol ether 3 (9 mmol) through a syringe, and the reaction mixture was stirred under nitrogen at 100 °C for 8 h. The resulting reaction mixture was treated with 5%  $NaHCO_3$  (20 mL), and the product was extracted with ether (25 mL  $\times$  3). After drying over anhydrous  $MgSO_4$ , the solvent was evaporated in vacuo. The crude product was purified by flash column chromatography with dichloromethane as an eluent to afford 4.

**1-[(Dimethylamino)carbonyl]-3-(methoxycarbonyl)-4-(1,1-dimethyl-2-oxopropyl)-1,4-dihydropyridine (4i):** oil;  $^1H$  NMR ( $CDCl_3$ )  $\delta$  0.85 (s, 3 H), 0.95 (s, 3 H), 2.05 (s, 3 H), 2.83 (s, 6 H), 3.55 (s, 3 H), 3.69 (d, 1 H,  $J = 6$  Hz), 4.82 (dd, 1 H,  $J = 6, 7$  Hz), 6.55 (d, 1 H,  $J = 7$  Hz), 7.60 (s, 1 H). Anal. Calcd for  $C_{15}H_{22}O_5N_2$ : C, 61.21; H, 7.53; N, 9.52. Found: C, 61.06; H, 7.69; N, 9.24.

**1-[(Dimethylamino)carbonyl]-3-(methoxycarbonyl)-4-[1-(methoxycarbonyl)benzyl]-1,4-dihydropyridine (4j).** The initially eluted diastereomer of 4j from column chromatography ( $CH_2Cl_2$ ): oil;  $^1H$  NMR ( $CDCl_3$ )  $\delta$  2.50 (s, 6 H), 3.68 (s, 3 H), 3.78 (s, 3 H), 4.01 (d, 1 H,  $J = 4$  Hz), 4.25 (t, 1 H,  $J = 4$  Hz), 5.27 (dd, 1 H,  $J = 4, 8$  Hz), 6.37 (dd, 1 H,  $J = 8, 1.5$  Hz), 7.0–7.25 (m, 5 H), 7.41 (d, 1 H,  $J = 1.5$  Hz). Anal. Calcd for  $C_{18}H_{22}O_5N_2$ : C, 63.68; H, 6.19; N, 7.82. Found: C, 63.80; H, 6.13; N, 8.00. Another diastereomer: oil;  $^1H$  NMR ( $CDCl_3$ )  $\delta$  2.92 (s, 6 H), 3.54 (s, 3 H), 3.63 (s, 3 H), 3.61–3.80 (m, 1 H), 3.98 (t, 1 H,  $J = 6$  Hz), 4.99 (dd, 1 H,  $J = 6, 8$  Hz), 6.53 (dd, 1 H,  $J = 8, 1.5$  Hz), 6.92–7.5 (m, 5 H), 7.69 (d, 1 H,  $J = 1.5$  Hz).

**1-[(Dimethylamino)carbonyl]-3-(methoxycarbonyl)-4-[1-(methoxycarbonyl)-1-methylethyl]-1,4-dihydropyridine (4k):** mp 79–80.5 °C;  $^1H$  NMR ( $CDCl_3$ )  $\delta$  1.08 (s, 3 H), 1.11 (s, 3 H), 2.97 (s, 6 H), 3.66 (s, 3 H), 3.72 (s, 3 H), 3.84 (d, 1 H,  $J = 6$  Hz), 4.99 (dd, 1 H,  $J = 6, 8$  Hz), 6.71 (d, 1 H,  $J = 8$  Hz), 7.76 (s, 1 H). Anal. Calcd for  $C_{15}H_{22}O_5N_2$ : C, 58.05; H, 7.14; N, 9.03. Found: C, 58.09; H, 7.05; N, 9.18.

**Synthesis of 1,3-Bis(methoxycarbonyl)-4-methyl-1,4-dihydropyridine or -4-*p*-tolyl-1,4-dihydropyridine (4l or 4m) with Methyl- or *p*-Tolylcopper. General Procedure.** Methyl- or *p*-tolyl lithium (13.5 mmol) was added to a suspension of CuI (2.57 g, 13.5 mmol) in 30 mL of THF at –78 °C. The mixture was stirred for 10 min and warmed to –30 °C and stirred for 20 min at that temperature. The mixture was transferred through a double-ended needle to a suspension of nicotinium salt in 15 mL of THF at –30 °C, which was prepared from methyl nicotinate (1.00 g, 7.3 mmol) and methyl chloroformate (0.64 mL, 8 mmol). The resulting mixture was stirred for 5–6 h at room temperature, then treated with 5%  $NaHCO_3$  (20 mL). The product was extracted with ether (25 mL  $\times$  3). After drying over anhydrous  $MgSO_4$ , the solvent was evaporated in vacuo. The crude product was purified by flash column chromatography with hexane and ethyl acetate (5:1) as eluent to afford 4l or 4m.

**1,3-Bis(methoxycarbonyl)-4-methyl-1,4-dihydropyridine (4l):**  $^1H$  NMR ( $CDCl_3$ )  $\delta$  1.16 (d, 3 H,  $J = 6.6$  Hz), 3.31 (dq, 1 H,  $J = 6.6, 4.6$  Hz), 3.75 (s, 3 H), 3.87 (s, 3 H), 5.08 (dd, 1 H,  $J = 4.6, 8$  Hz), 6.72 (dd, 1 H,  $J = 8, 1.5$  Hz), 7.87 (d, 1 H,  $J = 1.5$  Hz).

**1,3-Bis(methoxycarbonyl)-4-*p*-tolyl-1,4-dihydropyridine (4m):** mp 133–135 °C;  $^1H$  NMR ( $CDCl_3$ )  $\delta$  2.30 (s, 3 H), 3.63 (s, 3 H), 3.91 (s, 3 H), 4.41 (d, 1 H,  $J = 4.3$  Hz), 5.13 (dd, 1 H,  $J = 4.3, 8$  Hz), 6.85 (br d, 1 H,  $J = 8$  Hz), 8.03 (br s, 1 H). Anal. Calcd for  $C_{16}H_{17}O_5N$ : C, 66.89; H, 5.96; N, 4.88. Found: C, 66.57; H, 5.91; N, 5.18.

**Reaction of 3,4-Dichlorothiophene 1,1-Dioxide with 4-[2-(Alkoxy)alkyl]-Substituted 1,4-Dihydropyridines**

(4c,d,j,k) or 4-Alkyl- or 4-Aryl-Substituted 1,4-Dihydropyridines (4l,m). **General Procedure.** A mixture of a 1,4-dihydropyridine 4 (1 mmol) and 5 equiv of 7 was dissolved in 1 mL of toluene and heated to reflux. The reaction was monitored by TLC and stopped when a small amount of the secondary product appeared on TLC. The mixture was separated by flash column chromatography (dichloromethane or ethyl acetate-hexane) to afford 8 (8c,d,j-m).

**6,7-Dichloro-1-(ethoxycarbonyl)-3-(methoxycarbonyl)-4-[1-(methoxycarbonyl)benzyl]-1,4,4a,8a-tetrahydroquinoline (8c).** Diastereomerically pure 4c (initially eluted isomer) gave a single product with 7: mp 109–112 °C;  $^1H$  NMR ( $CDCl_3$ )  $\delta$  1.35 (t, 3 H,  $J = 6$  Hz,  $CO_2CH_2CH_3$ ), 2.92 (ddd, 1 H,  $J = 1.1, 2.5, 7$  Hz, H4a), 3.53–3.63 (m, 2 H, H4 and H9), 3.61 (s, 3 H,  $CO_2Me$ ), 3.78 (s, 3 H,  $CO_2Me$ ), 3.87 (ddd, 1 H,  $J = 1, 6, 7$  Hz, H8a), 4.25 (q, 2 H,  $J = 6$  Hz,  $CO_2CH_2Me$ ), 5.57 (dd, 1 H,  $J = 1, 2.5$  Hz, H5), 6.35 (d, 1 H,  $J = 6$  Hz, H8), 7.33 (s, 5 H, Ph), 8.20 (s, 1 H, H2). Anal. Calcd for  $C_{23}H_{23}NO_6Cl_2$ : C, 57.51; H, 4.83; N, 2.92. Found: C, 57.25; H, 5.10; N, 2.82.

**6,7-Dichloro-1-(ethoxycarbonyl)-3-(methoxycarbonyl)-4-[1-(methoxycarbonyl)-1-methylethyl]-1,4,4a,8a-tetrahydroquinoline (8d):** oil;  $^1H$  NMR ( $CDCl_3$ )  $\delta$  1.18 (s, 3 H), 1.22 (s, 3 H), 1.37 (t, 3 H,  $J = 7$  Hz), 3.10–3.38 (m, 2 H), 3.65 (s, 3 H), 3.75 (s, 3 H), 4.10–4.40 (m, 3 H), 5.50 (br s, 1 H), 6.46 (d, 1 H,  $J = 6$  Hz), 8.25 (s, 1 H).

**6,7-Dichloro-1-[(dimethylamino)carbonyl]-3-(methoxycarbonyl)-4-[1-(methoxycarbonyl)benzyl]-1,4,4a,8a-tetrahydroquinoline (8j).** Diastereomerically pure 4j (initially eluted isomer) gave a single product with 7: mp 172.5–175 °C;  $^1H$  NMR ( $CDCl_3$ )  $\delta$  2.89 (s, 6 H), 3.35–3.90 (m, 4 H), 3.67 (s, 3 H), 3.78 (s, 3 H), 5.70–5.78 (m, 1 H), 6.16 (d, 1 H,  $J = 7.5$  Hz), 7.32 (s, 5 H), 7.66 (s, 1 H). Anal. Calcd for  $C_{23}H_{24}O_5N_2Cl_2$ : C, 57.63; H, 5.05; N, 5.84. Found: C, 57.83; H, 5.33; N, 5.63.

**6,7-Dichloro-1-[(dimethylamino)carbonyl]-3-(methoxycarbonyl)-4-[1-(methoxycarbonyl)-1-methylethyl]-1,4,4a,8a-tetrahydroquinoline (8k):** mp 137.5–139 °C;  $^1H$  NMR ( $CDCl_3$ )  $\delta$  1.19 (s, 6 H), 2.90 (s, 6 H), 2.90–3.20 (m, 2 H), 3.68 (s, 3 H), 3.71 (s, 3 H), 4.22 (dd, 1 H,  $J = 7, 7$  Hz), 5.65 (br s, 1 H), 6.28 (d, 1 H,  $J = 7$  Hz), 7.71 (s, 1 H). Anal. Calcd for  $C_{19}H_{24}N_2O_5Cl_2$ : C, 52.99; H, 5.57; N, 6.34. Found: C, 52.91; H, 5.61; N, 6.50.

**2,3,5-Trichloro-7-[(dimethylamino)carbonyl]-9-(methoxycarbonyl)-10-[1-(methoxycarbonyl)-1-methylethyl]-6a,7,10,10a-tetrahydrobenzo[*f*]quinoline (9k):**  $^1H$  NMR ( $CDCl_3$ )  $\delta$  1.24 (s, 6 H), 2.61 (s, 6 H), 2.85–2.90 (m, 2 H), 3.64 (s, 3 H), 3.70 (s, 3 H), 4.39 (dd, 1 H,  $J = 6, 5$  Hz), 6.44 (d, 1 H,  $J = 6$  Hz), 7.27–7.35 (m, 2 H), 7.61 (s, 1 H).

**2,3,5-Trichloro-10-[(dimethylamino)carbonyl]-8-(methoxycarbonyl)-7-[1-(methoxycarbonyl)-1-methylethyl]-6a,7,10,10a-tetrahydrobenzo[*h*]quinoline (10k):**  $^1H$  NMR ( $CDCl_3$ )  $\delta$  1.20 (s, 6 H), 2.67 (s, 6 H), 2.93–3.20 (m, 2 H), 3.64 (s, 3 H), 3.68 (s, 3 H), 4.59 (d, 1 H,  $J = 4$  Hz), 5.65 (br s, 1 H), 7.45 (s, 1 H), 7.53 (br s, 2 H).

**6,7-Dichloro-1,3-bis(methoxycarbonyl)-4-methyl-1,4,4a,8a-tetrahydroquinoline (8l):** oil;  $^1H$  NMR ( $CDCl_3$ )  $\delta$  1.18 (d, 3 H,  $J = 7$  Hz), 2.26–2.90 (m, 2 H), 3.73 (s, 3 H), 3.84 (s, 3 H), 4.79 (dd, 1 H,  $J = 4, 6.5$  Hz), 6.10 (d, 1 H,  $J = 4$  Hz), 6.17 (d, 1 H,  $J = 5$  Hz), 8.61 (s, 1 H).

**6,7-Dichloro-1,3-bis(methoxycarbonyl)-4-(*p*-methylphenyl)-1,4,4a,8a-tetrahydroquinoline (8m):**  $^1H$  NMR ( $CDCl_3$ )  $\delta$  2.31 (s, 3 H), 2.63 (m, 1 H), 3.52 (s, 3 H), 3.70 (d, 1 H,  $J = 8$  Hz), 3.87 (s, 3 H), 4.82 (dd, 1 H,  $J = 3.7, 6.8$  Hz), 5.94 (d, 1 H,  $J = 5.5$  Hz), 6.02 (d, 1 H,  $J = 3.7$  Hz), 6.83–7.15 (m, 4 H), 8.20 (s, 1 H).

**Reaction of 3,4-Dichlorothiophene 1,1-Dioxide with 3-(Methoxycarbonyl)-4-(2-oxoalkyl)-Substituted 1,4-Dihydropyridines 4a,b,e-g,i in the Presence of 2,6-Lutidine. General Procedure.** To a solution of the 1,4-dihydropyridine (1 mmol) and 10 equiv of 7 in 2 mL of toluene was added 15 equiv of 2,6-lutidine. The mixture was heated to reflux. The reaction was monitored by TLC and stopped when a small amount of secondary product (9, 10) appeared on TLC. The resulting mixture was treated with 5%  $NaHCO_3$  (20 mL), and the product was extracted with dichloromethane (25 mL  $\times$  3). After drying over anhydrous  $MgSO_4$ , the solvent was evaporated in vacuo. The product was separated with flash column chromatography (1:2 or 1:4 ethyl acetate/hexane or dichloromethane).

**6,7-Dichloro-1-(ethoxycarbonyl)-3-(methoxycarbonyl)-4-(2-oxo-2-phenylethyl)-1,4,4a,8a-tetrahydroquinoline (8a):** mp 164.4–165.5 °C;  $^1\text{H NMR}$  ( $\text{CDCl}_3$ )  $\delta$  1.36 (t, 3 H,  $J = 7$  Hz), 2.9–3.56 (m, 4 H), 3.71 (s, 3 H), 4.31 (q, 2 H,  $J = 7$  Hz), 4.56 (dd, 1 H,  $J = 6, 6$  Hz), 5.87 (d, 1 H,  $J = 4$  Hz), 6.25 (d, 1 H,  $J = 6$  Hz), 7.3–7.6 (m, 3 H), 7.9–8.03 (m, 2 H), 8.15 (s, 1 H). Anal. Calcd for  $\text{C}_{22}\text{H}_{21}\text{O}_5\text{NCl}_2$ : C, 58.67; H, 4.70; N, 3.11. Found: C, 58.69; H, 4.41; N, 2.85.

**6,7-Dichloro-1-(ethoxycarbonyl)-3-(methoxycarbonyl)-4-[2-oxo-2-(*p*-methoxyphenyl)ethyl]-1,4,4a,8a-tetrahydroquinoline (8b):** oil;  $^1\text{H NMR}$  ( $\text{CDCl}_3$ )  $\delta$  1.36 (t, 3 H,  $J = 7$  Hz), 2.83–3.50 (m, 4 H), 3.72 (s, 3 H), 3.87 (s, 3 H), 4.28 (q, 2 H,  $J = 7$  Hz), 4.55 (dd, 1 H,  $J = 5, 5$  Hz), 5.86 (d, 1 H,  $J = 4$  Hz), 6.28 (d, 1 H,  $J = 5$  Hz), 6.95 (d, 2 H,  $J = 9$  Hz), 7.96 (d, 2 H,  $J = 9$  Hz), 8.27 (s, 1 H).

**6,7-Dichloro-1,3-bis(methoxycarbonyl)-4-(1-methyl-2-oxobutyl)-1,4,4a,8a-tetrahydroquinoline (8e).** Diastereomerically pure **4e** (initially eluted isomer) gave a single product with 7: oil;  $^1\text{H NMR}$  ( $\text{CDCl}_3$ )  $\delta$  0.96 (d, 3 H,  $J = 7$  Hz), 1.06 (t, 3 H,  $J = 7$  Hz), 2.35–3.30 (m, 5 H), 3.77 (s, 3 H), 3.88 (s, 3 H), 4.75 (dd, 1 H,  $J = 7, 4$  Hz), 5.82 (d, 1 H,  $J = 5$  Hz), 6.10 (d, 1 H,  $J = 4$  Hz), 8.14 (s, 1 H). Anal. Calcd for  $\text{C}_{18}\text{H}_{21}\text{O}_5\text{NCl}_2$ : C, 53.78; H, 2.98; N, 5.29. Found: C, 53.74; H, 3.48; N, 5.26.

**6,7-Dichloro-1,3-bis(methoxycarbonyl)-4-(1-methyl-2-oxo-2-phenylethyl)-1,4,4a,8a-tetrahydroquinoline (8f).** Diastereomerically pure **4f** (initially eluted isomer) gave a single

product with 7: oil;  $^1\text{H NMR}$  ( $\text{CDCl}_3$ )  $\delta$  1.06 (d, 3 H,  $J = 7$  Hz), 2.76 (dd, 1 H,  $J = 7, 7$  Hz), 3.20 (dd, 1 H,  $J = 7, 4$  Hz), 3.83 (s, 3 H), 3.88 (s, 3 H), 4.4–4.7 (m, 1 H), 4.85 (dd, 1 H,  $J = 7, 3$  Hz), 5.80 (d, 1 H,  $J = 6$  Hz), 5.86 (d, 1 H,  $J = 3$  Hz), 7.3–7.7 (m, 3 H), 8.0–8.3 (m, 3 H).

**6,7-Dichloro-1,3-bis(methoxycarbonyl)-4-(1,1-dimethyl-2-oxopropyl)-1,4,4a,8a-tetrahydroquinoline (8g):** oil;  $^1\text{H NMR}$  ( $\text{CDCl}_3$ )  $\delta$  1.13 (s, 6 H), 2.22 (s, 3 H), 3.10–3.31 (m, 2 H), 3.75 (s, 3 H), 3.84 (s, 3 H), 4.40 (dd, 1 H,  $J = 6, 6$  Hz), 5.55 (br s, 1 H), 6.46 (d, 1 H,  $J = 6$  Hz), 8.27 (s, 1 H).

**6,7-Dichloro-1-[(dimethylamino)carbonyl]-3-(methoxycarbonyl)-4-(1,1-dimethyl-2-oxopropyl)-1,4,4a,8a-tetrahydroquinoline (8i):** oil;  $^1\text{H NMR}$  ( $\text{CDCl}_3$ )  $\delta$  1.10 (s, 3 H), 1.15 (s, 3 H), 2.22 (s, 3 H), 2.92 (s, 6 H), 2.90–3.25 (m, 2 H), 3.71 (s, 3 H), 4.25 (ddd, 1 H,  $J = 6, 6, 1.5$  Hz), 5.70 (dd, 1 H,  $J = 2, 1.5$  Hz), 6.28 (d, 1 H,  $J = 6$  Hz), 7.27 (s, 1 H).

**Acknowledgment.** We are grateful to Chisso, Ltd. for the gift of silyl chlorides and to the Grant-in-Aid for Special Project Research (59104002) administered by the Japanese Ministry of Education, Science, and Culture.

**Supplementary Material Available:** IR spectral data for **4a–k,m** and **8a–c,e,i–l** and mass spectral data for **4b–k,m** and **8a–c,e,j–l** (3 pages). Ordering information is given on any current masthead page.

## Electrophilic Sulfur Transfer Reactions in Organic Synthesis. Preparation of a Diastereomer of the Key Macrocyclic Component of Griseoviridin

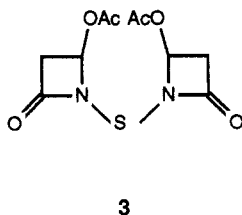
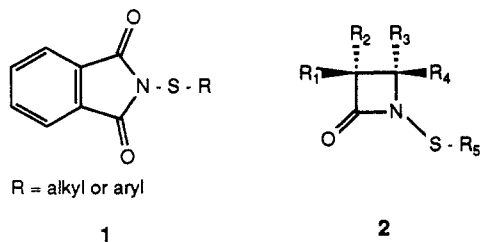
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Received July 10, 1986

The utility of electrophilic sulfur transfer reactions was demonstrated by the synthesis of a diastereomer of the key macrocyclic cysteine containing component of griseoviridin. The key step involved direct reaction at the sulfur of *N*-(carbobenzyloxy)-*S*-phthalimido-*L*-cysteine *tert*-butyl ester with the anion derived from methyl 3-oxa-5(*S*)-[(*tert*-butyldimethylsilyl)oxy]hexanoate.

*N*-(Alkylthio)- or *N*-(arylthio)phthalimides **1** serves as sulfonyl transfer reagents upon reaction with a variety of heteroatoms<sup>1</sup> and active methylene compounds.<sup>2</sup> We have found this type of electrophilic reaction at sulfur to be especially useful for the preparation of several sulfur-containing molecules of biological interest, including the novel *N*-*S* containing  $\beta$ -lactams **2** and **3**.<sup>3</sup> Although re-



lated attempts to sulfonylate active methylene compounds often gave mixtures of mono- and bisulfonylated products, we have recently found that anions derived from active methylene compounds **4** react with *S*-phthalimidocysteine derivatives **5** to give the corresponding monosulfonylated products **6** in good yield (eq 1). Since these reactions provide an effective way of transferring an entire cysteine unit to another carbon framework by reaction at sulfur, no racemization of the  $\alpha$ -chiral center is expected. This has encouraged us to study the utility of cysteine-based sulfur transfer reactions for the synthesis of other biologically interesting molecules. Herein we report on the successful use of a cysteine-based electrophilic sulfur transfer reaction as the key step in the synthesis of a diastereomer of a primary component (**7**) of griseoviridin (**8**).

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<sup>\*</sup> Recipient of an NIH Research Career Development Award, 1983–1988.